# ASCOCHALASIN, A NEW CYTOCHALASIN FROM ASCOCHYTA HETEROMORPHA

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ABSTRACT.—The structure of a new cytochalasin, ascochalasin, isolated together with deoxaphomin from *Ascochyta heteromorpha*, pathogen of oleander (*Nerium oleander*), has been determined mainly by <sup>1</sup>H-nmr and <sup>13</sup>C-nmr spectroscopy.

A previous analysis of the organic culture filtrate extracts of Ascochyta heteromorpha (Sch. et Sacc.) Curzi, a pathogenic fungus isolated from oleander (*Nerium oleander* L.), led to the isolation of the known cytochalasins A [1] and B [2] (1). A further investigation of the extracts has revealed the presence of another two cytochalasins: deoxaphomin [3], isolated from *Phoma* sp. (strain S 298) (2) and identified as a biosynthetic precursor of cytochalasin B (3), and a new cytochalasin, which we have named ascochalasin [5].

In the present note we report the isolation and the structure elucidation of ascochalasin.

The chromatographic purification of culture filtrate organic extracts was performed using two SiO<sub>2</sub> columns in two successive steps. Cytochalasins A and B were obtained together with a mixture of products with chromatographic behavior very close to that of **2**. The mixture was purified by preparative tlc first on SiO<sub>2</sub> and then on reversed-phase plates, affording pure deoxaphomin and ascochalasin. Deoxaphomin [**3**] was identified by comparing our spectral data, ir, uv, nmr, and ms, with those reported by Binder and Tamm (2). The acetyl derivative **4** of **3** gave a <sup>1</sup>H-nmr spectrum in which the appear-



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ance of two singlets at  $\delta$  2.04 and 1.94 due to two acetyl groups was noteworthy. In addition, the cims (isobutane as reagent gas) of 4 showed a pseudomolecular ion at m/z [MH]<sup>+</sup> 548 (5%); other significant peaks were present at m/z [MH – CH<sub>2</sub>CO]<sup>+</sup> 506, [MH – HOAc]<sup>+</sup> 488, [MH – HOAc – CH<sub>2</sub>CO]<sup>+</sup> 446, [MH – 2×HOAc]<sup>+</sup> 428, and [MH – C<sub>7</sub>H<sub>7</sub> – HOAc – CH<sub>3</sub>CO]<sup>+</sup> 354. These findings were in full agreement with structure 4 as 7,20-diacetyldeoxaphomin.

Ascochalasin [5] was an amorphous solid and had the molecular formula  $C_{29}H_{39}NO_4$  from hrms (m/z [M]<sup>+</sup> 465.2873, calcd m/z 465.2869). Other significant peaks were detected at m/z 463.2715 ( $C_{29}H_{37}NO_4$ ), likely formed by loss of  $H_2$  in the source from the molecular ion (465.2873); 372.2191 ( $C_{22}H_{30}NO_4$ ) and 354.2075 ( $C_{22}H_{28}NO_3$ ) originated from the ion at m/z 463.2715 by loss of a benzyl group and  $H_2O$ , respectively; the ion at m/z 270.1496 ( $C_{17}H_{20}NO_2$ ) arose from the fragment at m/z 354.2075 by losses of  $H_2O$  and a cyclopentadienyl moiety. This fragmentation pattern is coincident with that of deoxaphomin [3] except for the loss of two hydrogens.

The structure of **5** was assigned mainly on the basis of spectral evidence arising from <sup>1</sup>H nmr, including a series of decoupling experiments, and <sup>13</sup>C-nmr data. Tables 1 and 2 list the proton and carbon shifts, respectively. The <sup>1</sup>H-nmr spectrum of **5** resulted in very close approximation to that of deoxaphomin, recorded under the same experimen-

Proton	Compounds		
	5	3	
Н-3	3.30 ddd	3.30 ddd	
H-4	3.14 dd	3.13 dd	
H-5	2.88 dq	2.84 dq	
H-7	3.94 d	3.97 d	
H-8	2.48 dd	2.47 dd	
H-10	2.74 dd	2.69 dd	
H-10'	2.50 dd	2.47 dd	
3H-11	1.06 d	1.03 d	
H-12	5.32 br s	5.29 br s	
H-12'	5.12 br s	5.13 br s	
H-13	6.20 ddd	6.18 ddd	
H-14	5.42 ddd	5.37 ddd	
H-15	2.15 ddd	2.08 ddd	
H-15′			
H-16			
2H-17	0.85-2.00	0.85-2.00	
2H-18			
2H-19			
H-20	3.30 m	4.26 m	
H-21	0.85-2.00	6.70 dd	
H-22	0.85-2.00	7.12 dd	
Me-C(16)	0.94 d	0.90 d	
2',6'			
3',5'	7.20	7.20	
4'			

 TABLE 1.
 <sup>1</sup>H-nmr Data of Ascochalasin [5] and Deoxaphomin [3].<sup>a,b</sup>

<sup>a</sup>J (Hz) **3**, **5**:  $J_{3,4} = 3.3$ ,  $J_{3,10} = 5.5$ ,  $J_{3,10'} = 3.3$ ,  $J_{4,5} = 5.9$ ,  $J_{5,11} = 6.6$ ,  $J_{7,8} = 10.3$ ,  $J_{8,13} = 9.6$ ,  $J_{10,10'} = 13.6$ ,  $J_{13,14} = 15.4$ ,  $J_{13,15} = 1.8$ ,  $J_{14,15} = 9.6$ ,  $J_{14,15'} = 3.3$ ,  $J_{15,15'} = 12.3$ ,  $J_{16,Me} = 6.6$ , **3**:  $J_{20,21} = 8.8$ ,  $J_{20,22} = 1.8$ ,  $J_{21,22} = 15.4$ .

<sup>b</sup>Chemical shifts are in  $\delta$  values (ppm) from TMS.

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Carbon	Compound 5	Carbon	Compound 5
Carbon           C-1            C-3            C-4            C-5            C-6            C-7            C-8            C-9            C-10            C-11            C-12            C-13	164.1s 53.2d 45.8d 31.9d 143.9s 70.6d 50.7d 65.7s 44.3t 13.6q 114.1t 127.4d	C-17          C-18          C-19          C-20          C-21          C-22          C-23          C-1'          C-1'          C-1'          C-2'          C-4'	34.7 t 25.2 t 41.5 t 70.3 d 39.2 t 29.6 t 191.6 s 22.2 q 129.3 s 129.1 d 128.9 d 127.0 d
C-14 C-15 C-16	137.4 d 29.6 t 33.8 d	C-5' C-6'	128.9 d 129.1 d

TABLE 2. <sup>13</sup>C-nmr Spectral Data of Ascochalasin [5].<sup>a,b</sup>

<sup>a</sup>Assignment made also by the comparison with the spectral data of cytochalasin B [2] (4) and cytochalasin K [7] (5).

<sup>b</sup>Chemical shifts are in  $\delta$  values (ppm) from TMS.

tal conditions, although several differences were observed due to the lack of a double bond in 5 in comparison with 3. In fact, an accurate examination of the <sup>1</sup>H-nmr spectrum of 5 revealed the absence of two double doublets at  $\delta$  7.12 and 6.70, corresponding to the H-22 and H-21 of 3. Therefore, 5 has only two double bonds located between C-13 and C-14 ( $J_{13,14} = 15.4$  Hz, *trans*-oriented) and C-6 and C-12 (exocyclic double bond). The abovementioned structural difference is supported also from the upfield shift ( $\Delta \delta 0.96$ ) in 5 at  $\delta$  3.30 of H-20 which appeared as a more complex system. In addition, the region of the aliphatic protons ( $\delta 0.85$ -2.00) showed an increase of multiplets, which should be attributed to the C-21 and C-22 methylene groups.

The <sup>13</sup>C-nmr data listed in Table 2 further support the structure assigned to ascochalasin. In particular, as compared with the data published for cytochalasin B [2] (4), the spectrum of **5** showed the absence of the carbon signals present in 2 at  $\delta$  154.2 and 119.2, attributed to C-21 and C-22, respectively, while two signals appeared in **5** at  $\delta$  39.2 and 29.6, due to C-21 and C-22, respectively, in addition to the other signals of aliphatic methylene groups. These latter signals were recorded at  $\delta$  41.5, 34.7, and 25.2 corresponding to C-19, C-17, and C-18, respectively. The C-15, resonating at the same frequency of C-22, gave rise to a single intense signal at  $\delta$  29.6. Moreover, the examination of the chemical shifts of carbonyl groups corroborated the structure of **5**. In fact, the amidic carbonyl group (O=C-1) afforded a signal at  $\delta$  164.1, while the other carbonyl group (O=C-23) resonated at  $\delta$  191.6, which is a typical chemical shift value for ketonic groups and consistent with the shift reported for the same carbon ( $\delta$ 196.9) in cytochalasin K [7] (5). The ketonic nature of C-23 in **5** determined, in comparison with **2**, the upfield shift of C-9 ( $\Delta \delta$  18.7) which appeared at  $\delta$  65.7. A similar chemical shift value is observed for the C-9 ( $\delta$  63.0) of **7** (5).

The saturated nature of the ketone in **5** was supported by its ir data analyzed in comparison to those of **3**. In fact the band corresponding to the double bond conjugated with the ketone which appeared at  $1622 \text{ cm}^{-1}$  in **3** was absent in **5**. Moreover, a broad intense band at  $1715 \text{ cm}^{-1}$  due to both ketonic and lactam carbonyl groups appeared in the ir spectrum of **5**, while these groups showed in **3** as two absorptions in the 1710- $1670 \text{ cm}^{-1}$  region.

On standing in pyridine and  $Ac_2O$ , 5 formed a diacetyl derivative 6. Its cims

showed a peak at m/z 548 formed from the pseudomolecular ion [MH]<sup>+</sup> by loss of two hydrogen atoms. Significant peaks, formed from the ion at m/z 548, were observed at m/z [MH – H<sub>2</sub> – HOAc]<sup>+</sup> 488, [MH – H<sub>2</sub> – HOAc – CH<sub>2</sub>CO]<sup>+</sup> 446, and [MH – H<sub>2</sub> –  $2 \times \text{HOAc}$ ]<sup>+</sup> 428.

The <sup>1</sup>H-nmr spectrum of **6** showed, in comparison to that of **5**, the downfield shift of H-7 and H-20 ( $\Delta \delta 0.89$  and 0.39, respectively) which appeared at  $\delta 4.83$  and 3.69, respectively, in **6**. In addition, the two singlets corresponding to the two acetyl groups appeared at  $\delta 2.06$  and 1.94.

In conclusion, the structure of ascochalasin [5] can be formulated as (7S, 13E, 16R, 20R)-7,20-dihydroxy-16-methyl-10-phenyl-[13] cytochalasan-6(12),13-diene-1,23-dione.

The very interesting biological properties of cytochalasins (6,7) suggest the use of ascochalasin in a number of bioassays (phytotoxic, mycotoxic, and antibacterial activity) in comparison with other cytochalasins and some of their derivatives.

## EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Ir spectra were recorded on Perkin-Elmer 399 instrument in  $CH_2Cl_2$  solutions; uv spectra were measured on a Varian-Cary 210 spectrophotometer in MeOH solutions; <sup>1</sup>H-nmr and <sup>13</sup>C-nmr spectra were recorded in CDCl<sub>3</sub> at 270 and 67.88 MHz, respectively, on a Bruker spectrometer; ei and ci (isobutane as reagent gas) mass spectra were recorded at 70 and 250 eV, respectively, on a Kratos Ltd. 80 mass spectrometer. Analytical and preparative tlc were performed on SiO<sub>2</sub> (20 × 20 cm, Merck Kieselgel 60 F<sub>254</sub>, 0.25 mm) and on reversed-phase plates (20 × 20 cm, Whatman Stratocrom C-18 0.2 mm) plates; the spots were visualized by exposure to uv light and/or by spraying the plates first with 10% H<sub>2</sub>SO<sub>4</sub> in MeOH and then with 3% phosphomolybdic acid in MeOH, followed by heating for 5 min at 105°. Cc was carried out on SiO<sub>2</sub> (Merck, Kieselgel 60 0.063–0.2 mm).

FUNGUS SPECIES.—A. heteromorpha was isolated in 1985 from oleander (Nerium oleander) grown in a nursery near Bari, and deposited in the fungus collection of the Istituto Tossine e Micotossine da Parassiti Vegetali del CNR, Bari, Italy.

GROWTH OF FUNGUS.—Single spore cultures of *A. beteromorpha*, freshly reisolated from infected oleander plants and maintained on slants of potato-dextrose-agar, were used. The fungus was cultured in 1000-ml Roux flasks containing 200 ml of a semisynthetic liquid medium (8). The cultures were incubated at 25° for 21 days and then filtered.

EXTRACTION AND ISOLATION OF CYTOCHALASINS.—The culture filtrates (9.0 liters, pH 7) were lyophilized, resuspended in distilled H<sub>2</sub>O (650 ml), and extracted with CH<sub>2</sub>Cl<sub>2</sub>(5 × 600 ml). The organic extracts were combined, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated under reduced pressure. The residue (600 mg) was chromatographed on a SiO<sub>2</sub> column eluting with CHCl<sub>3</sub>-iPrOH (9:1). The first fraction yielded crude cytochalasin A [1] (57 mg) the successive eluate contained small amounts of cytochalasin A, cytochalasin B [2], and a mixture of products (324 mg) with a chromatographic behavior very similar to that of 2. This mixture was further purified on a SiO<sub>2</sub> column eluting with CHCl<sub>3</sub>-iPrOH (93:7). Crude cytochalasin B (70 mg) was obtained and a fraction consisting of small amounts of cytochalasin B and a mixture of two products (82 mg) with  $R_f$  lower than 2. The latter was further purified on SiO<sub>2</sub> plates (CHCl<sub>3</sub>-iPrOH, 93:7) affording crude cytochalasin B (38 mg) and a crude mixture of two other products (20 mg). The mixture was finally purified on reversed-phase plates (MeCN-H<sub>2</sub>O, 6:4), giving two homogeneous products with  $R_f$  0.50 (ascochalasin, 5 mg) and  $R_f$  0.45 (deoxaphomin, 7 mg).

Ascochalasin [5].—Ascochalasin was obtained as an amorphous solid: ir  $\nu \max 3580-3500$ , 3400, 1715, 1605 cm<sup>-1</sup>; uv  $\lambda \max \min (\log \epsilon) \le 220$ ; <sup>1</sup>H nmr and <sup>13</sup>C nmr see Tables 1 and 2, respectively; ms *m/z* (rel. int.) 465 (3), 463 (5), 372 (10), 354 (22), 270 (14), 211 (70), 91 (100).

Descaphomin [3].—Descaphomin was obtained as an amorphous solid: ir  $\nu \max 3590-3540$ , 3410, 1710–1670, 1622, 1600 cm<sup>-1</sup>; uv  $\lambda \max nm (\log \epsilon)$  215 (4.03) and 237 (3.87); <sup>1</sup>H nmr see Table 1; ms m/z (rel. int.) 463 (8), 445 (15), 372 (30), 354 (20), 174 (18), 91 (100).

Diacetylascochalasin [6].—Ascochalasin (6 mg) was acetylated with Ac<sub>2</sub>O/pyridine. The usual workup of the reaction afforded an oil, which was purified on SiO<sub>2</sub> plates (eluent CHCl<sub>3</sub>-MeOH, 98:2) giving pure 6 (5.6 mg, 80%): uv  $\lambda$  max nm (log  $\epsilon$ ) <220; <sup>1</sup>H-nmr spectrum was very similar to that of 5 except for the downfield shifts of H-7 and H-20 ( $\Delta \delta 0.89$  and 0.39, respectively) which appeared at  $\delta$  4.83 and 3.69, respectively, in 6, and the presence of two new signals (both singlets) at  $\delta$  2.06 and 1.94, corresponding to the two acetyl groups on C-7 and C-20; ms m/z (rel. int.)  $[MH-H_2]^+$  548 (3),  $[MH-H_2-HOAc]^+$  488 (8),  $[MH-H_2-HOAc-CH_2CO]^+$  446 (11),  $[MH-H_2-2 \times HOAc]^+$  428 (100).

Diacetyldeoxaphomin [4].—Deoxaphomin (4 mg) was acetylated using the same method to convert 5 to 6. The pure product (4.6 mg, 82%) showed uv  $\lambda$  max (log  $\epsilon$ ) 220 (4.2), 240 (3.5); <sup>1</sup>H nmr was very similar to that of 3 except for the presence of the two new signals (both singlets at  $\delta$  2.04 and 1.94) corresponding to two acetyl groups on C-7 and C-20; ms m/z (rel. int.) [MH]<sup>+</sup> 548 (5), [MH – CH<sub>2</sub>CO]<sup>+</sup> 506 (10), [MH – HOAc]<sup>+</sup> 488 (65), [MH – HOAc – CH<sub>2</sub>CO]<sup>+</sup> 446 (55), [MH – 2 × HOAc]<sup>+</sup> 428 (100), [MH – C<sub>7</sub>H<sub>7</sub> – HOAc – CH<sub>3</sub>CO]<sup>+</sup> 354 (10), 336 (8), 174 (5), 91 (10).

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